Natural Sciences Chemistry Final

Author: Briana Hamilton

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4. Chapter: Chemistry Final						
. Chemistry Fin	al Questions					
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4.1.1. The part of the atom that is not in the valence electrons

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The part of the atom that is not in the valence electrons

Kernal

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4.1.2. Part of the atom involved in bonding

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Part of the atom involved in bonding

Valence Shell Electrons

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4.1.3. Predicts the distribution of the electrons in energy levels

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Predicts the distribution of the electrons in energy levels

Aufbau Principle

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4.1.4. The number of protons in the nucleus

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The number of protons in the nucleus

Atomic Number

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4.1.5. This subatomic particle is not found in the nucleus

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This subatomic particle is not found in the nucleus

Electron

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4.1.6. The average mass distribution of the isotopes of an element

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The average mass distribution of the isotopes of an element

Atomic Mass

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4.1.7. Occurs when electrons are shared

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Occurs when electrons are shared

Double Bond

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4.1.8. Occurs when both electrons in the shared pair come from the same atom

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Occurs when both electrons in the shared pair come from the same atom

Coordinate Covalent Bond

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4.1.9. A reaction that produces a precipitate or a gas

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A reaction that produces a precipitate or a gas

Non reversible Reaction

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4.1.10. Occurs when electrons are transferred

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Occurs when electrons are transferred

Ionic

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A group ofcovalently bonded atoms that react as a single charged unit

Polyatomic Ion

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4.1.12. Sum of atomic weights of a compound

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Sum of atomic weights of a compound

Molecular Weight

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4.1.13. Energy necessary for a reactant to yield a product

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Energy necessary for a reactant to yield a product

Activation Energy

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4.1.14. When a substance can act as an acid or base

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When a substance can act as an acid or base

Amphoteric

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4.1.15. Neutralize HCI

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Neutralize HCI

Antacid

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4.1.16. When an ionic compound breaks down

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When an ionic compound breaks down

Dissocation

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4.1.17. poles in a battery anode and cathode

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poles in a battery anode and cathode

• Electrode

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4.1.18. Cathode

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Cathode

Electroplating

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4.1.19. Required energy; produces its own energy

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Required energy; produces its own energy

Electrolytic Cell

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4.1.20. Balancing oxidation and reduction

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Balancing oxidation and reduction

Half Reaction

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Heat absorbed or released when a reaction takes place at a constant pressure

Heat of Reaction

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4.1.22. Exist between nonpolar no dipole gas molecules such as H^2

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Exist between nonpolar no dipole gas molecules such as H^2

London Dispersion

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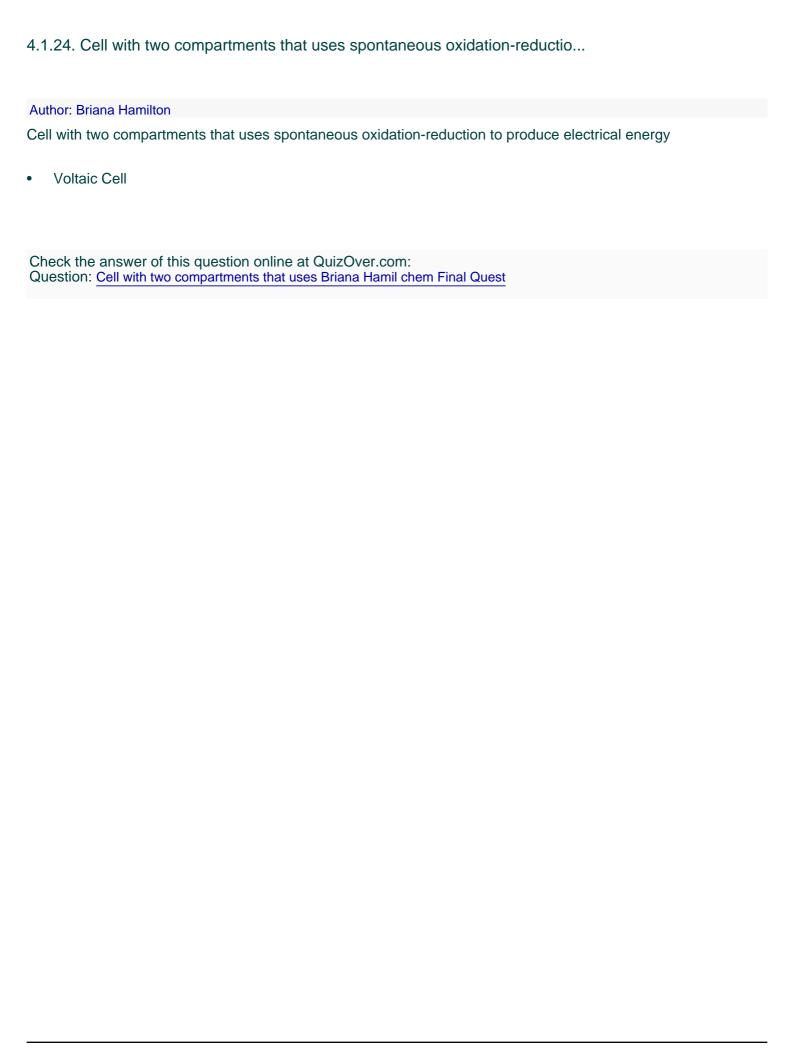
4.1.23. Weak due to synchronization

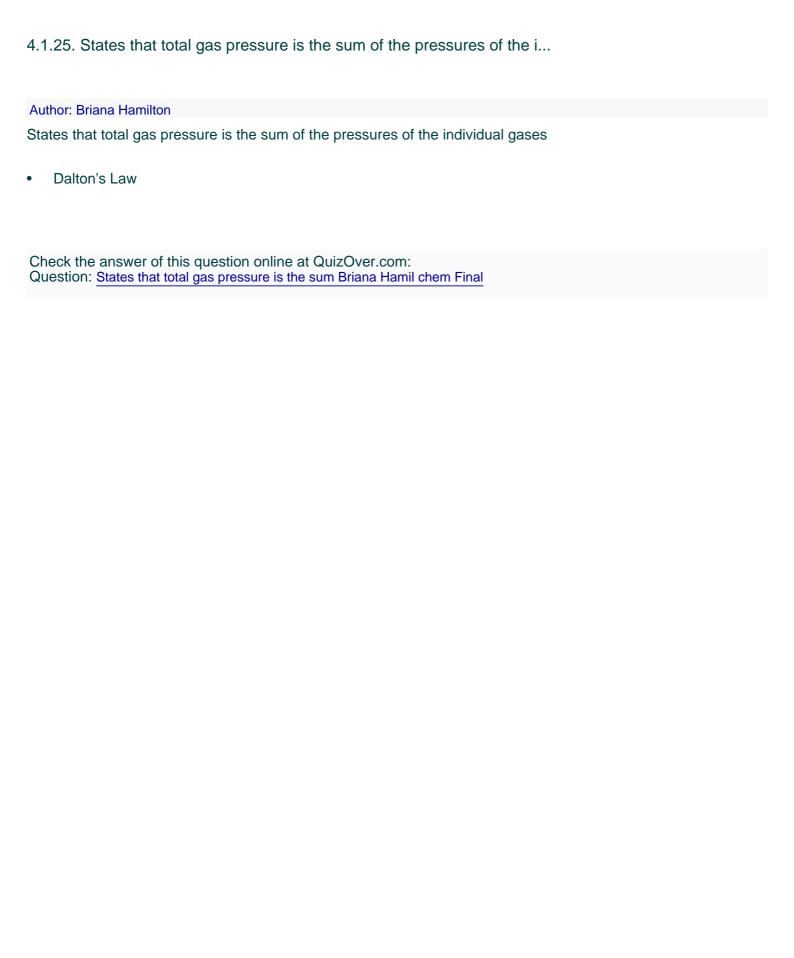
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Weak due to synchronization

London Dispersion

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Describes how much of a solid can be dissolved at a given temperature

Solubility

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4.1.27. A solid solution of two or more metals

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A solid solution of two or more metals

Alloy

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4.1.28. Particles too big to pass through a filter but settle upon standing

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Particles too big to pass through a filter but settle upon standing

Suspension

Check the answer of this question online at QuizOver.com: Question: Particles too big to pass through a filter Briana Hamil chem Final

4.1.29. A solid with water attached

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A solid with water attached

Anhydrous

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4.1.30. The pressure of a gas on the surface of a liquid

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The pressure of a gas on the surface of a liquid

Vapor Pressure

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